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CHAPTER 5 Periodic Classification of Elements

Modern Periodic Table → Atomic number of an element is a more fundamental property than its atomic mass. → According to the Modern Periodic law : The properties of elements are a periodic function of their atomic number. → All the anomalies of Mendeleev's

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classification disappear.

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Need for classification of elements:.

Increase in the discovery of different elements made it difficult to organise...

2. Dobereiner's triads:.
Johann Wolfgang Dobereiner, a German chemist, classified the known elements in ...

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formula of a molecule made up of three atoms of oxygen. (2012 OD)

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means identifying similar species and grouping them together. 2. Lavoisier divided elements into two main types ...

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Chapter 5 - Periodic Classification of Elements Exercise 85 Solution 1 K is in group 1. Therefore, the oxide will be K_2O .

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Video Timeline: 0:00:00 Starts 0:08:17
Introduction 0:11:34 Classification of

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Periodic Table- 1) Dobereiner's Triads
0:19:40 2) Newland's Law of Octaves
0:26...

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